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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.003 Å R factor = 0.056 wR factor = 0.167 Data-to-parameter ratio = 13.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A 1:1 cocrystal of pyridine and 4-hydroxybenzoic acid

The title cocrystal, $C_5H_5N \cdot C_7H_6O_3$, consists of pyridine and 4hydroxybenzoic acid molecules which are linked by $O-H \cdots N$ and $O-H \cdots O$ hydrogen bonds, leading to zigzag chains along the *b* axis. Received 3 September 2006 Accepted 12 September 2006

Comment

In recent years, hydrogen bonds have attracted the interest of chemists and have been widely used to design and synthesize one-, two- and three-dimensional supramolecular compounds (Aakeröy *et al.*, 2002). 4-Hydroxybenzoic acid is a good hydrogen-bond donor and can form cocrystals with other organic molecules (Vishweshwar *et al.*, 2003). Here we report the crystal structure of the title cocrystal, (I).



Compound (I) consists of pyridine and 4-hydroxybenzoic acid neutral molecules (Fig. 1 and Table 1). 4-Hydroxybenzoic acid forms $O-H\cdots N$ hydrogen bonds with pyridine and $O-H\cdots O$ hydrogen bonds with two neighboring 4-hydroxybenzoic acid molecules, forming zigzag chains along the *b* axis (Fig. 2).

Experimental

All reagents were commercially available and of analytical grade. 4-Hydroxybenzoic acid (2.0 mmol, 0.28 g) was dissolved in pyridine (12 ml). The mixture was stirred for 30 min at room temperature and was then filtered. Colorless crystals were obtained after 8 d by slow evaporation of the filtrate.

	Crystal data	
	$C_{5}H_{5}N \cdot C_{7}H_{6}O_{3}$ $M_{r} = 217.22$ Monoclinic, $P_{2_{1}}/c$ $a = 12.163 (5) \text{ Å}$ $b = 9.002 (4) \text{ Å}$ $c = 11.171 (5) \text{ Å}$ $\beta = 117.34 (1)^{\circ}$ $V = 1086.6 (8) \text{ Å}^{3}$	Z = 4 $D_x = 1.328 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 0.10 \text{ mm}^{-1}$ T = 298 (2) K Block, colorless $0.30 \times 0.20 \times 0.20 \text{ mm}$
у	Data collection Bruker SMART APEX CCD diffractometer ω scans Absorption correction: none 6841 measured reflections	2118 independent reflections 1602 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.174$ $\theta_{\text{max}} = 26.0^{\circ}$

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Figure 1

The asymmetric unit of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level.



Figure 2

The crystal structure of (I). Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.057$ $wR(F^2) = 0.167$ S = 1.002118 reflections 153 parameters H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0853P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.20 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.32 \text{ e } \text{\AA}^{-3}$

Tab	le 1	
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Selected geometric parameters (Å, $^\circ).$

C7-O1	1.228 (2)	C7-O2	1.297 (2)
O1-C7-C1	122.53 (15)	O2-C7-C1	115.63 (14)

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O2-HO2\cdots N1^{i}$	0.843 (10)	1.753 (11)	2.592 (2)	173 (3)
O3−HO3···O1 ⁱⁱ	0.825 (10)	1.868 (11)	2.686 (2)	171 (2)

The value of $R_{\rm int}$ is very large because the crystal is not stable at room temperature and effloresces easily. Therefore, there were some difficulties in collecting the data. H atoms of the hydroxy and carboxyl groups were located in difference maps, and O–H lengths were restrained to 0.82 (1) Å. Other H atoms were positioned geometrically, with C–H = 0.93 Å, and were refined as riding with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT-Plus* (Bruker, 2001); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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